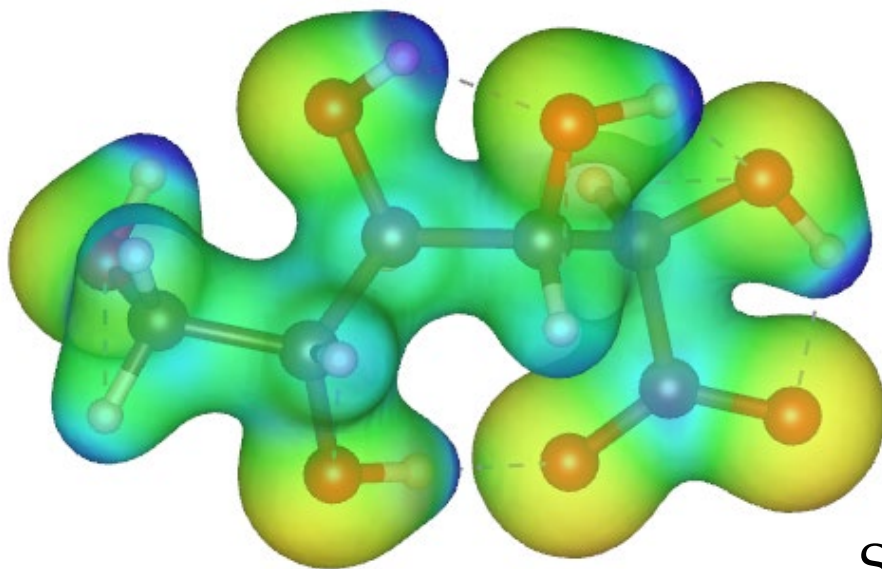


Computational Chemistry:

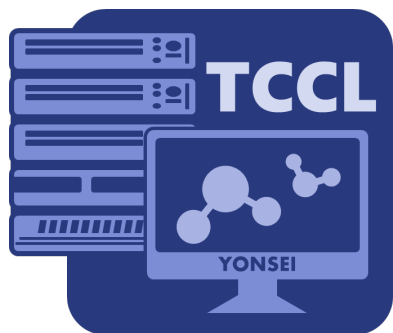
Hands-on Electronic Structure Calculations



Youngsam Kim

Suwon, April 26 - 28, 2023

대한화학회 제131회 학술발표회



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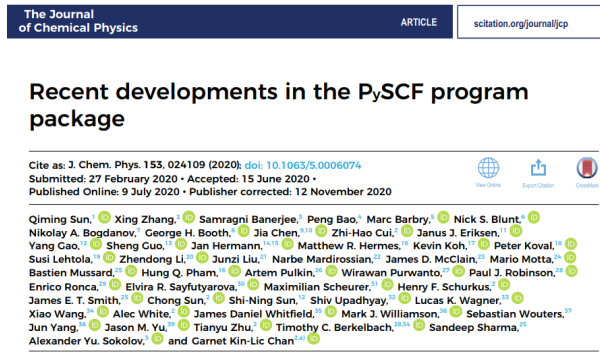
Introduction: PySCF

Software Focus

PySCF: the Python-based simulations of chemistry framework



Qiming Sun,^{1*} Timothy C. Berkelbach,² Nick S. Blunt,^{3,4} George H. Booth,⁵ Sheng Guo,^{1,6} Zhendong Li,¹ Junzi Liu,⁷ James D. McClain,^{1,6} Elvira R. Sayfutyarova,^{1,6} Sandeep Sharma,⁸ Sebastian Wouters⁹ and Garnet Kin-Lic Chan^{1*}



Python-based Simulations of Chemistry Framework

- Open-source
- Python-based (intuitive)
- Gaussian type orbital(GTO) for cluster systems
- GTO + planewave for periodic systems

Introduction: Settings

① Google Colab & 새 노트 click

The screenshot shows the Google Colab interface. The top bar displays the Colab logo and the text "Colaboratory에 오신 것을 환영합니다" (Welcome to Colaboratory). Below this, there are navigation options: "파일" (File), "수정" (Edit), "보기" (View), "삽입" (Insert), "런타임" (Runtime), "도구" (Tools), and "도움말" (Help). The left sidebar contains a "목차" (Table of Contents) section with options like "시작하기" (Get started), "데이터 과학" (Data science), "머신러닝" (Machine learning), "추가 리소스" (Additional resources), and "색션" (Sections). The main content area shows a list of notebooks under the "최근 사용" (Recently Used) tab. The list includes a notebook titled "Colaboratory에 오신 것을 환영합니다" (Welcome to Colaboratory) and several "Untitled" notebooks. A red circle with the number "1" and a red arrow points to the "새 노트" (New Notebook) button at the bottom right of the interface.

예	최근 사용	Google Drive	GitHub	업로드
	노트 필터링			
	제목	마지막 연 시간	처음 연 시간	
	Colaboratory에 오신 것을 환영합니다	오후 7:13	4월 12일	
	Untitled1.ipynb	오후 7:13	오후 7:13	
	대한화학회tutorial.ipynb	4월 13일	4월 13일	
	Untitled0.ipynb	4월 12일	4월 12일	
				새 노트 취소

Introduction: Settings

- ② Define/Modify note title
- ③ PySCF installation → `pip install pyscf`
- ④ Add 코드

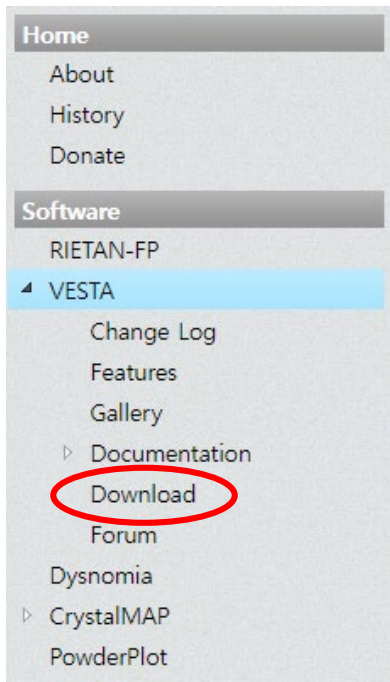
② Left-click + modification

③ Take care of the spacing, run ► button (or Ctrl + Enter)

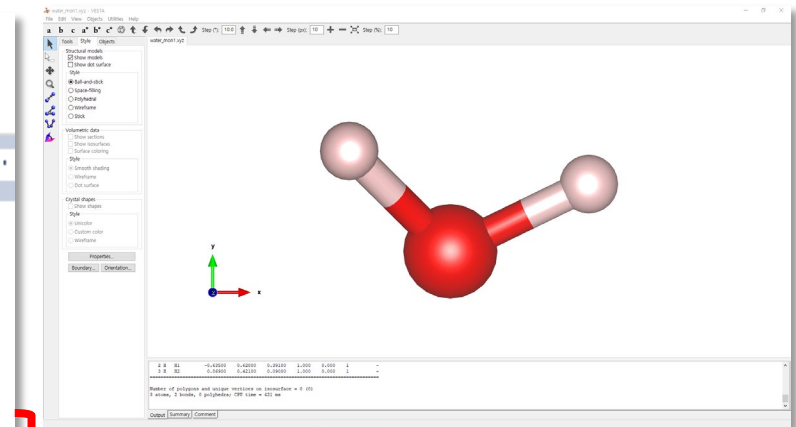
The screenshot shows a Jupyter Notebook titled "대한화학회 튜토리얼 part 1.ipynb". The interface includes a top bar with navigation icons and a right sidebar with system status (RAM, 디스크). The main area contains a code cell with the command `pip install pyscf`. The output shows the installation progress, including the download of `pyscf-2.2.1-cp39-cp39-manylinux_2_17_x86_64.manylinux2014_x86_64.whl` (47.7 MB) and the successful installation of `pyscf-2.2.1`. Red arrows and callouts highlight the title, the code cell, and the run button.

Introduction: Settings

⑤ VESTA (Visualization for Electronic and Structural Analysis) (optional)



```
water_mon1.xyz ...
파일  흥  보기
1 2 3 4
3 ← Num. of atoms
str ← Comment line
O 0.000 0.000 0.000
H -0.635 0.628 0.391
H 0.869 0.421 0.090
```



Element X Y Z

워드패드(or 메모장) → save *.xyz

Part I. Basic Density Functional Theory Calculation

Input & Output for H₂O monomer

Exercise) H₂O dimer interaction energy

Part II. Geometry Optimization

Input & Output for LiH *database

Exercise) CH₃OH optimization

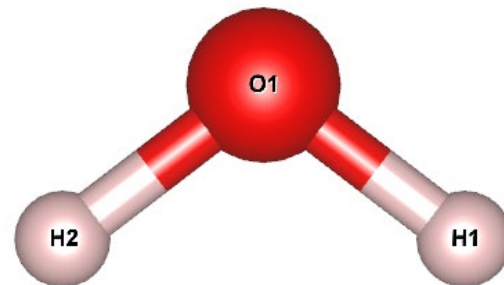
Part III. Dispersion Correction

Part I. Basic DFT Calculation

<Input code>

```
from pyscf import gto, dft

mol = gto.M(
    atom='''
    0 0 0 0
    H 0 0.77 0.58
    H 0 -0.77 0.58''',
    spin=0,
    charge=0,
    basis='ccpvdz',
    verbose=4) ①
```



```
mydft = dft.UKS(mol)
mydft.xc = 'b3lyp'
mydft.kernel()

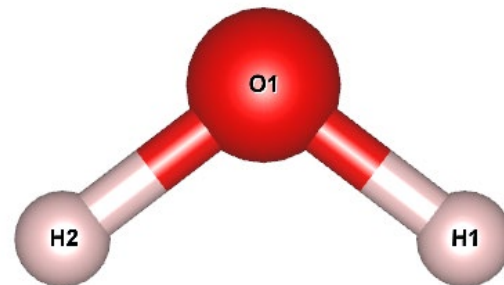
mydft.analyze() ②
```

!! Take caution !!

bracket (), spacing,
comma, enter ...

Run →  Click or Ctrl+Enter

Part I. Basic DFT Calculation



<Input code>

```
from pyscf import gto, dft
```

Import package

```
mol = gto.M(
    atom='''
    0 0 0 0
    H 0 0.77 0.58
    H 0 -0.77 0.58''',
    spin=0,
    charge=0,
    basis='ccpvdz',
    verbose=4)
```

Define 'mol' from 'gto'

→ atom : nuc. type, coordinates (xyz, Å)

→ spin, charge: spin state(2S) and charge

→ basis : basis sets ⇒ Prof. Sherrill's note

* <http://vergil.chemistry.gatech.edu/notes/>

→ Verbose : print level(4)

```
mydft = dft.UKS(mol)
mydft.xc = 'b3lyp'
mydft.kernel()

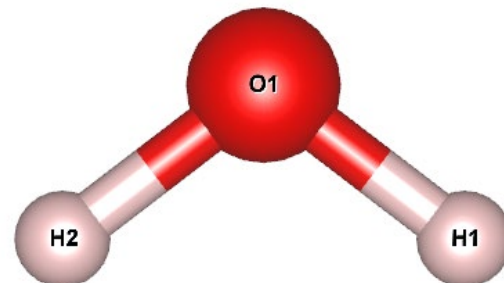
mydft.analyze()
```

→ (UKS)RKS : (Un-)Restricted Kohn Sham
w/o α, β distinction. orbital occupation = 2 or 0

→ kernel() : perform calculation

→ analyze() : population analyze

Part I. Basic DFT Calculation



<Output>

```
System: uname_result(system='Linux', node='4149422508eb', release='5.10.147+', version='#1 SMP Sat Dec 10 16:00:40 UTC 2022', machine='x86_64') Threads 2
Python 3.9.16 (main, Dec 7 2022, 01:11:51)
[GCC 9.4.0]
numpy 1.22.4  scipy 1.10.1
Date: Tue Apr 25 03:55:53 2023
PySCF version 2.2.1
PySCF path /usr/local/lib/python3.9/dist-packages/pyscf
```

```
[CONFIG] conf_file None
[INPUT] verbose = 4
[INPUT] num. atoms = 3
[INPUT] num. electrons = 10
[INPUT] charge = 0
[INPUT] spin (= nelec alpha-beta = 2S) = 0
[INPUT] symmetry False subgroup None
[INPUT] Mole.unit = angstrom
```

- Check if the input is **well defined as the user thinks/wants!**

[INPUT]	Symbol	X	Y	Z	unit	X	Y	Z	unit	Magmom
[INPUT]	1 O	0.000000000000	0.000000000000	0.000000000000	AA	0.000000000000	0.000000000000	0.000000000000	Bohr	0.0
[INPUT]	2 H	0.000000000000	0.770000000000	0.580000000000	AA	0.000000000000	1.455089115915	1.096041152248	Bohr	0.0
[INPUT]	3 H	0.000000000000	-0.770000000000	0.580000000000	AA	0.000000000000	-1.455089115915	1.096041152248	Bohr	0.0

```
nuclear repulsion = 9.12662691069514
number of shells = 11
number of NR pGTOs = 40
number of NR cGTOs = 24
basis = ccpvdz
ecp = {}
CPU time: 9.60
```

```
number electrons alpha = 5  beta = 5
XC library pyscf.dft.libxc version 6.1.0
  S. Lehtola, C. Steigemann, M. J.T. Oliveira, and M. A.L. Marques., SoftwareX 7, 1-5 (2018)
XC functionals = b3lyp
  P. A. M. Dirac., Math. Proc. Cambridge Philos. Soc. 26, 376 (1930)
  F. Bloch., Z. Phys. 57, 545 (1929)
  A. D. Becke., Phys. Rev. A 38, 3098 (1988)
  C. Lee, W. Yang, and R. G. Parr., Phys. Rev. B 37, 785 (1988)
  B. Miehlich, A. Savin, H. Stoll, and H. Preuss., Chem. Phys. Lett. 157, 200 (1989)
  S. H. Vosko, L. Wilk, and M. Nusair., Can. J. Phys. 58, 1200 (1980)
```

Part I. Basic DFT Calculation

<Input code> in Gaussian 16

```
#p
#bv5lyp/cc-pvdz iop(3/76=100002000) iop(3/77=0720008000) iop(3/78=0810010000)

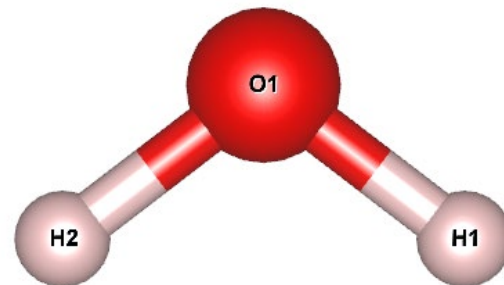
title

0 1
O 0. 0. 0.
H 0. 0.77 0.58
H 0. -0.77 0.58
```

<Input code> in ORCA

```
! B3LYP cc-pvdz
* xyz 0 1
O 0. 0. 0.
H 0. 0.77 0.58
H 0. -0.77 0.58
*
```

Part I. Basic DFT Calculation



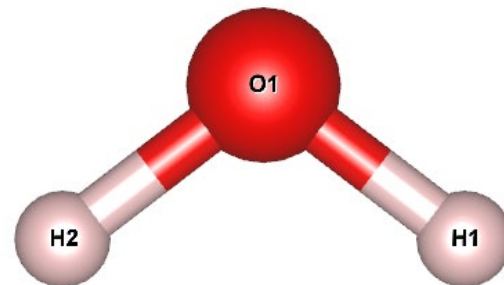
<Output>

```
init E= -76.3176487258378
  HOMO = -0.424214880461059  LUMO = 0.00576850782758609
cycle= 1 E= -76.2907860084321  delta_E= 0.0269  |g|= 0.711  |ddm|= 1.37
  HOMO = -0.115580962229854  LUMO = 0.0866957495862257
cycle= 2 E= -76.2059987829547  delta_E= 0.0848  |g|= 0.983  |ddm|= 0.973
  HOMO = -0.28894456934566  LUMO = 0.0500835157204298
cycle= 3 E= -76.3830990955477  delta_E= -0.177  |g|= 0.0234  |ddm|= 0.588
  HOMO = -0.283089328138573  LUMO = 0.054203061741246
cycle= 4 E= -76.3831985069832  delta_E= -9.94e-05  |g|= 0.00375  |ddm|= 0.0136
  HOMO = -0.283413378722867  LUMO = 0.0538996564196925
cycle= 5 E= -76.3832007801309  delta_E= -2.27e-06  |g|= 0.000462  |ddm|= 0.00203
  HOMO = -0.283555113081661  LUMO = 0.0538724659684704
cycle= 6 E= -76.3832008204877  delta_E= -4.04e-08  |g|= 7.88e-06  |ddm|= 0.000284
  HOMO = -0.283555920718238  LUMO = 0.0538710838505195
cycle= 7 E= -76.3832008205172  delta_E= -2.95e-11  |g|= 1.18e-06  |ddm|= 1.38e-05
  HOMO = -0.283555604005165  LUMO = 0.0538712791419524
Extra cycle E= -76.3832008205176  delta_E= -4.12e-13  |g|= 7.73e-07  |ddm|= 1.57e-06
converged SCF energy = -76.3832008205176
```

HOMO, LUMO → Gap

Total energy [E_h]

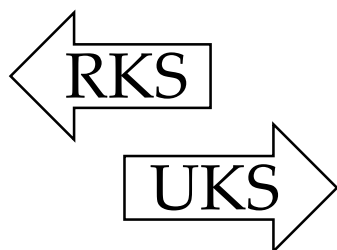
Part I. Basic DFT Calculation



<Output>

```
**** SCF Summaries ****
Total Energy = -76.383200820517644
Nuclear Repulsion Energy = 9.126626910695141
One-electron Energy = -123.103556146575698
Two-electron Coulomb Energy = 46.923017812384003
DFT Exchange-Correlation Energy = -9.329289397021089
```

```
**** MO energy ****
MO #1 energy= -19.1186491430855 occ= 2
MO #2 energy= -0.987661195417317 occ= 2
MO #3 energy= -0.510664437864799 occ= 2
MO #4 energy= -0.357633758735553 occ= 2
MO #5 energy= -0.283555604005165 occ= 2
MO #6 energy= 0.0538712791419524 occ= 0
MO #7 energy= 0.129377406916558 occ= 0
MO #8 energy= 0.565111003576567 occ= 0
MO #9 energy= 0.606639719197764 occ= 0
MO #10 energy= 0.903258789871577 occ= 0
MO #11 energy= 0.922489891876981 occ= 0
MO #12 energy= 0.997298355595242 occ= 0
MO #13 energy= 1.22165464571497 occ= 0
MO #14 energy= 1.23085906956994 occ= 0
MO #15 energy= 1.43785119825297 occ= 0
MO #16 energy= 1.60095726644132 occ= 0
MO #17 energy= 1.6599205573074 occ= 0
MO #18 energy= 2.11013769008181 occ= 0
MO #19 energy= 2.1413010108255 occ= 0
MO #20 energy= 2.91220076954952 occ= 0
MO #21 energy= 2.96986766410649 occ= 0
MO #22 energy= 3.12174633634056 occ= 0
MO #23 energy= 3.47733786606679 occ= 0
MO #24 energy= 3.72564591069997 occ= 0
```



$$* E_{tot} = E_{NN} + T + E_{Ne} + J + E_{XC}$$

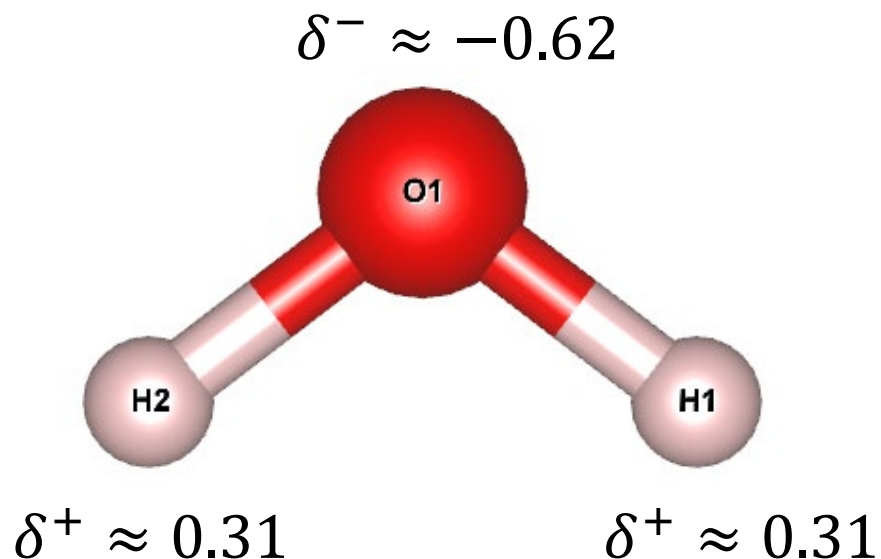
```
**** MO energy ****
```

	alpha	beta	alpha	beta
MO #1 energy=	-19.11864969637	-19.1186496461307	occ= 1	1
MO #2 energy=	-0.987661462457649	-0.987661372193143	occ= 1	1
MO #3 energy=	-0.510664614938504	-0.510664605726537	occ= 1	1
MO #4 energy=	-0.357633943450735	-0.357633831652488	occ= 1	1
MO #5 energy=	-0.283555801071953	-0.283555756823679	occ= 1	1
MO #6 energy=	0.0538711765905053	0.0538711791352494	occ= 0	0
MO #7 energy=	0.129377341036489	0.129377321194523	occ= 0	0
MO #8 energy=	0.565110929334846	0.565110851128378	occ= 0	0
MO #9 energy=	0.606639581274333	0.606639541955059	occ= 0	0
MO #10 energy=	0.903258653429998	0.90325869870356	occ= 0	0
MO #11 energy=	0.9224897322983	0.92248975994865	occ= 0	0
MO #12 energy=	0.997298267744648	0.997298291328731	occ= 0	0
MO #13 energy=	1.22165458103591	1.22165453335582	occ= 0	0
MO #14 energy=	1.23085895447495	1.2308588830455	occ= 0	0
MO #15 energy=	1.43785111432482	1.43785104947603	occ= 0	0
MO #16 energy=	1.60095713547502	1.60095715504632	occ= 0	0
MO #17 energy=	1.65992047448728	1.65992038796426	occ= 0	0
MO #18 energy=	2.11013752941266	2.11013749434837	occ= 0	0
MO #19 energy=	2.14130082015479	2.14130080322529	occ= 0	0
MO #20 energy=	2.91220059386519	2.91220065570533	occ= 0	0
MO #21 energy=	2.96986750213131	2.96986751186434	occ= 0	0
MO #22 energy=	3.1217461533732	3.12174623431986	occ= 0	0
MO #23 energy=	3.477337702458	3.47733770897637	occ= 0	0
MO #24 energy=	3.7256457306752	3.72564579502286	occ= 0	0

Part I. Basic DFT Calculation

<Output>

```
** Mulliken pop on meta-lowdin orthogonal AOs **
** Mulliken pop **
pop of 0 0 1s      1.99996
pop of 0 0 2s      1.64653
pop of 0 0 3s      0.00367
pop of 0 0 2px     1.99177
pop of 0 0 2py     1.29611
pop of 0 0 2pz     1.65644
pop of 0 0 3px     0.00228
pop of 0 0 3py     0.00940
pop of 0 0 3pz     0.00137
pop of 0 0 3dxy    0.00000
pop of 0 0 3dyz    0.00435
pop of 0 0 3dz^2   0.00159
pop of 0 0 3dxz    0.00126
pop of 0 0 3dx2-y2 0.00159
pop of 1 H 1s     0.67570
pop of 1 H 2s     0.01110
pop of 1 H 2px    0.00235
pop of 1 H 2py    0.00131
pop of 1 H 2pz    0.00139
pop of 2 H 1s     0.67570
pop of 2 H 2s     0.01110
pop of 2 H 2px    0.00235
pop of 2 H 2py    0.00131
pop of 2 H 2pz    0.00139
** Mulliken atomic charges **
charge of 00 =   -0.61632
charge of 1H =    0.30816
charge of 2H =    0.30816
```



B3LYP/ccpvdz

Population analysis

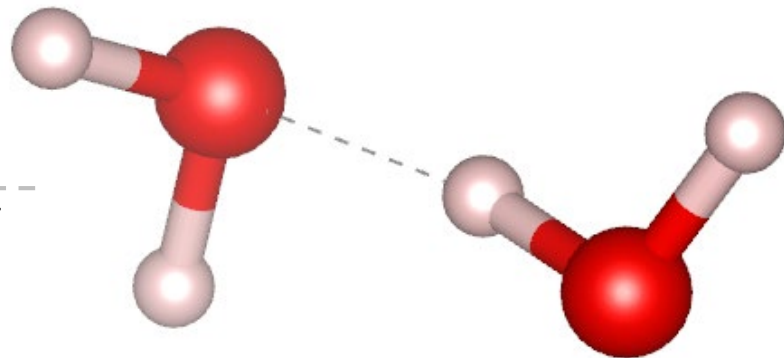
→ Mulliken charge (LCAO-MO)

→ Bader charge, ...

Part I. Basic DFT Calculation

Exercise 1. Water dimer interaction energy

	Ele.	X	Y	Z
Mon1	O	0.000	0.000	0.000
	H	-0.635	0.628	0.391
	H	0.869	0.421	0.090
<hr/>				
Mon2	O	-0.277	0.479	-2.567
	H	-0.975	-0.048	-2.956
	H	-0.230	0.212	-1.603



* Interaction energy

$$E(AB) - E(A) - E(B)$$

B3LYP/ccpvdz

Result) **-6.810 kcal/mol**

```
e = mydft.kernel()
```

```
e1 = mydft2.kernel()
```

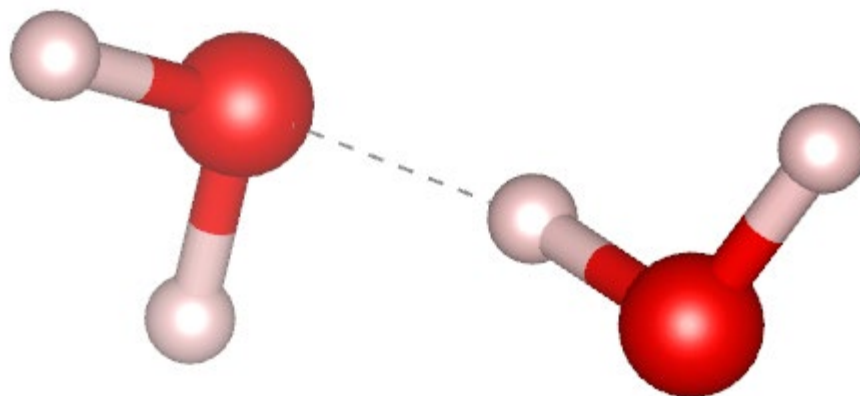
```
e2 = mydft3.kernel()
```

```
print((e - e1 - e2)*627.509)
```

* 1 Hartree = 627.509 kcal/mol

Part I. Basic DFT Calculation

Exercise 1. Water dimer interaction energy



	B3LYP	PBE	
cc-pVDZ	-6.810	-7.985	Ref. -3.395 kcal/mol
cc-pVTZ	-4.469	-5.565	DLPNO-CCSD(T)-F12/aug-cc-pVQZ
cc-pVQZ	-3.659	-4.757	

Part I. Basic DFT Calculation



PCCP

PAPER

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View Journal | View Issue

Check for updates

Cite this: *Phys. Chem. Chem. Phys.*,
2017, 19, 32184

A look at the density functional theory zoo with the advanced GMTKN55 database for general main group thermochemistry, kinetics and noncovalent interactions†

Lars Goerigk,^a Andreas Hansen,^b Christoph Bauer,^b Stephan Ehrlich,^b Asim Najibi^a and Stefan Grimme^{a,b}

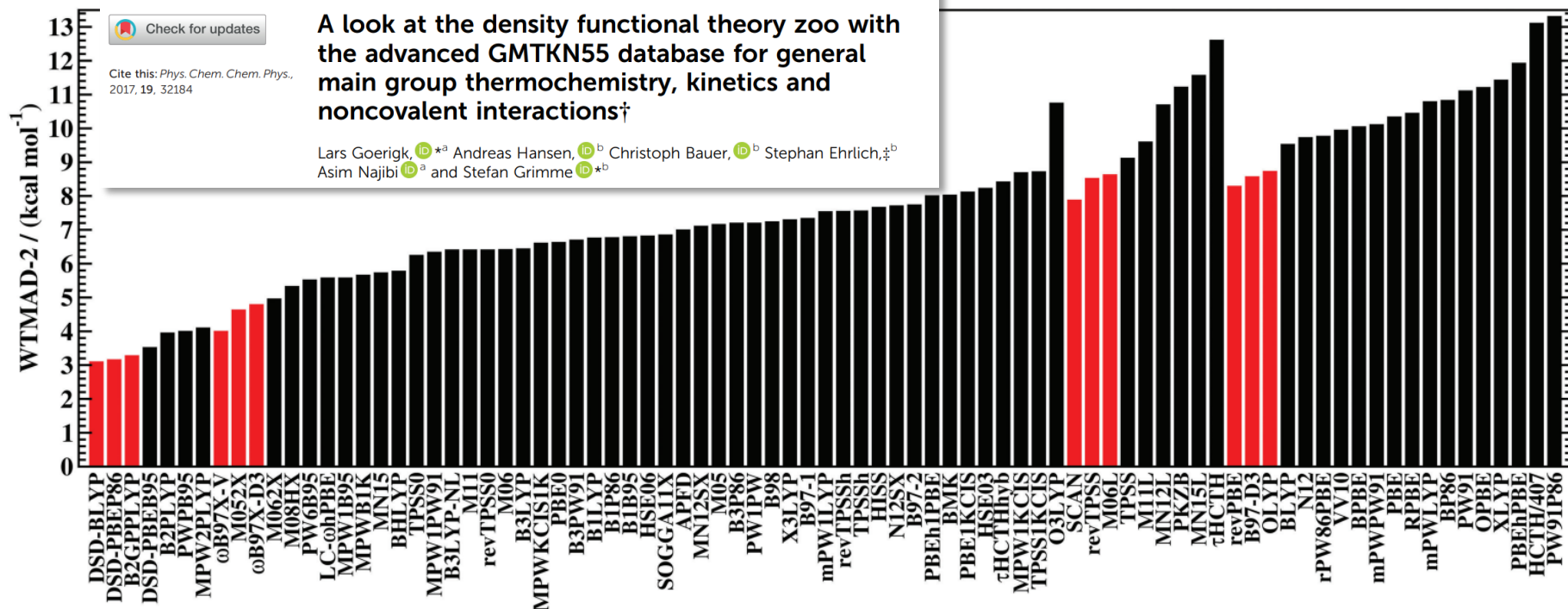


Fig. 8 Final WTMAD-2 values over the entire GMTKN55 for all assessed 83 dispersion-corrected DFAs (kcal mol⁻¹). The red bars indicate the three best approaches on their respective rung of Jacob's Ladder. The suffix "D3" was omitted in all cases, unless it is needed to avoid ambiguity.

- Comprehensive insight into {functional/basis sets} is required.

Part II. Geometry Optimization

** pip install pyberny (geometry optimizer)

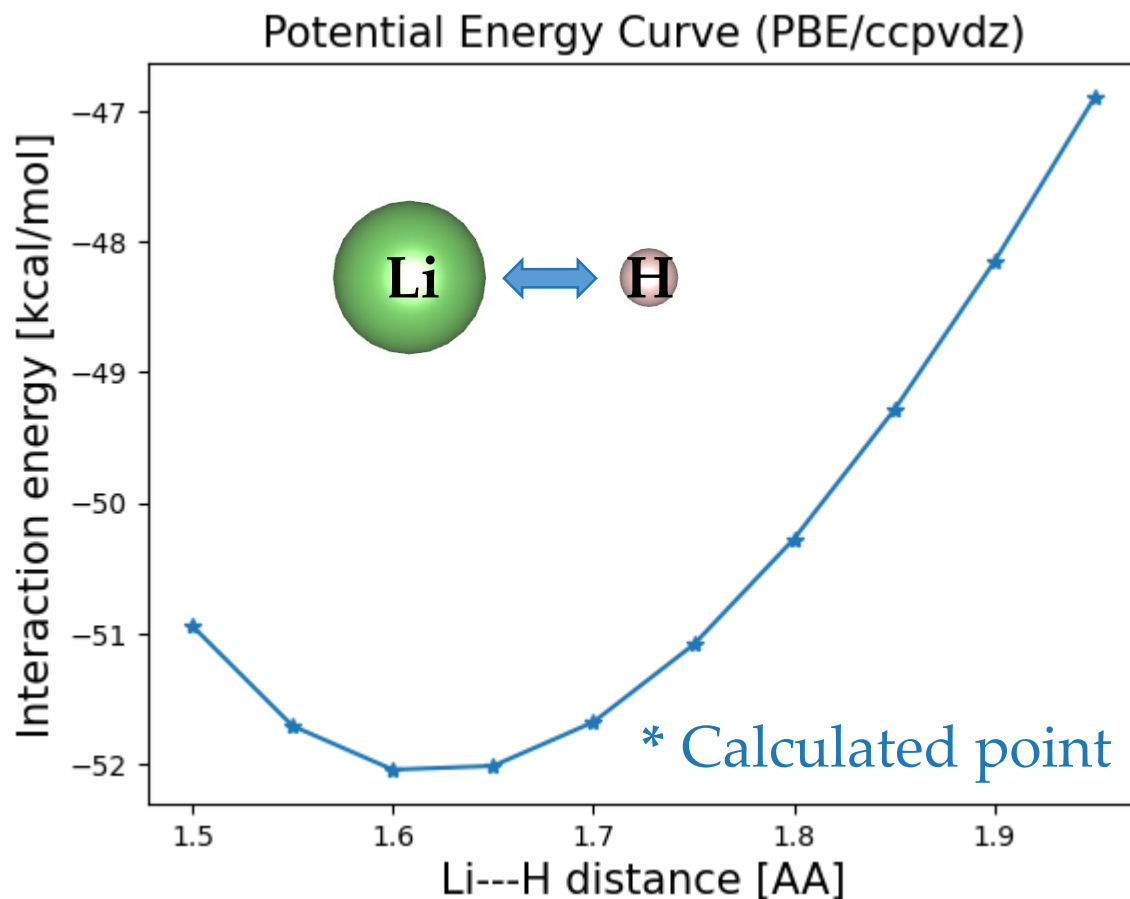
```
▶ pip install pyberny
```

```
Looking in indexes: https://pypi.org/simple, https://us-python.pkg.dev/colab-wheels/public/simple/  
Collecting pyberny  
  Downloading pyberny-0.6.3-py3-none-any.whl (27 kB)  
Requirement already satisfied: numpy<2.0,>=1.15 in /usr/local/lib/python3.9/dist-packages (from pyberny) (1.22.4)  
Installing collected packages: pyberny  
Successfully installed pyberny-0.6.3
```

<Input code>

```
from pyscf import gto, dft  
from pyscf.geomopt.berny_solver import optimize  
  
mol = gto.M(atom='Li 0 0 0; H 0 0 1.5', basis='ccpvdz', verbose = 4)  
mydft = dft.RKS(mol)  
mydft.xc = 'pbe'  
mol_eq = optimize(mydft)  
print(mol_eq.atom_coords()*0.529) # 1bohr = 0.529 Angstrom
```

Part II. Geometry Optimization



- Finding optimal(most stable) structure for a given system {charge, spin state, ...}

Part II. Geometry Optimization

<Output>

```
INFO:pyscf.geomopt.berny_solver.140709643640160:5 * All criteria matched
```

```
[[ 0.          0.         -0.06005129] →X Y Z (Li)
```

```
[ 0.          0.          1.55954897]] →X Y Z (H)
```

```
print(mol_eq.atom_coords()*0.529) # 1bohr = 0.529 Angstrom
```

⇒ Optimal distance Li-H (PBE/ccpvdz) : **1.62 Å**

** verbose = 4

Geometry optimization cycle 4

Cartesian coordinates (Angstrom)

Atom	New coordinates			dX	dY	dZ
Li	0.000000	0.000000	-0.058466	0.000000	0.000000	-0.008572
H	0.000000	0.000000	1.558466	0.000000	0.000000	0.008572

Part II. Geometry Optimization

Database <Cluster system>

 **Computational Chemistry Comparison and Benchmark DataBase** Release 22 (May 2022) Standard Reference Database 101 [National Institute of Standards and Technology](http://www.nist.gov)

Home Experimental Calculated Comparisons Resources FAQ Help

You are here: Home

The **CCCBDB** contains:

Experimental and computed (quantum mechanics) thermochemical data for a selected set of 2186 **gas-phase** atoms and small molecules.

Tools for comparing experimental and computational ideal-gas thermochemical properties.

Vibrational Frequencies, Rotational Constants, Electric Dipole, Electric Quadrupole, Polarizabilities

Molecules in the **CCCBDB** mostly have the following constraints:

- Well-established experimental heat of formation.
- Atoms with atomic number less than 36 (Krypton) with only a few transition metals. We have added a few molecules containing Te, I, and Xe
- Less than 15 heavy atoms and less than 30 atoms total. Except for a few larger molecules: tetracene, triphenylmethane, coronene, and C60.

Citation	NIST Computational Chemistry Comparison and Benchmark Database NIST Standard Reference Database Number 101 Release 22, May 2022, Editor: Russell D. Johnson III http://cccbdb.nist.gov/ DOI:10.18434/T47C7Z
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Please send questions, comments, corrections, additions and suggestions to cccbdb@nist.gov.



Part II. Geometry Optimization

Database <Cluster system>

 **Computational Chemistry Comparison and Benchmark DataBase** Release 22 (May 2022) Standard Reference Data

Home Experimental **Calculated** Comparisons Resources

You are here: Home

The CCCBDB contains:

Experimental and computed (quantum mechanics) thermochemical data for a wide variety of small molecules. Tools for comparing experimental and computational ideal-gas thermochemical data, including Vibrational Frequencies, Rotational Constants, Electric Dipole, etc.

Energy		
Geometry	Calculated geometry	AIM bond orders
Vibrations	Rotation	One type of bond
Electrostatics	Point group	
Entropy and Heat Capacity	State symmetry	
Reaction	<r2>	
Lookup by property	Z-matrix	
	Bad Calculations	

①

Calculated Geometries

Please enter the chemical formula

②

Part II. Geometry Optimization

Database <Cluster system>

Basis sets

Methods with standard basis sets

Methods

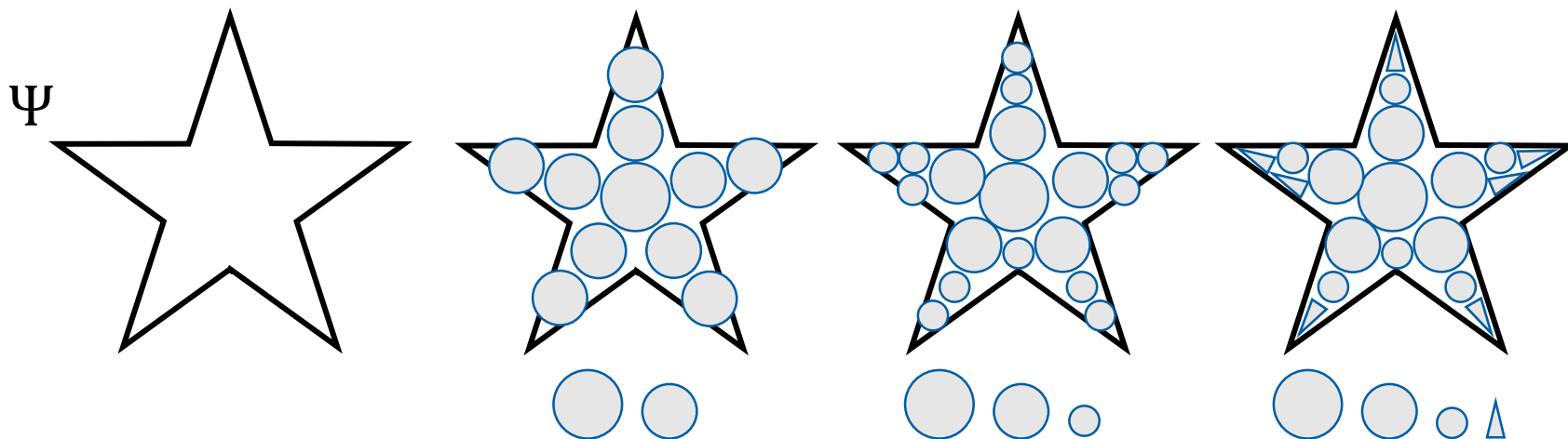
		STO-3G	3-21G	3-21G*	6-31G	6-31G*	6-31G**	6-31+G**	6-311G*	6-311G**	6-31G(2df,p)	6-311+G(3df,2p)	6-311+G(3df,2pd)	TZVP	cc-pVDZ	cc-pVTZ	
hartree fock	HF	1.511	1.640	1.640	1.640	1.636	1.630	1.629	1.608	1.607	1.628	1.608	1.608	1.612	1.619	1.608	
	LSDA	1.534	1.646	1.646	1.641	1.636	1.629	1.627	1.605	1.602	1.624	1.599	1.597	1.609	1.615	1.599	
density functional	BLYP	1.537	1.638	1.638	1.632	1.630	1.622	1.621	1.603	1.600	1.621	1.597	1.596	1.607	1.611	1.597	
	B1B95	1.530	1.636	1.636	1.633	1.628	dnf	1.621	1.599	1.599	1.621	1.597	1.596	1.605	1.611	1.596	
	B3LYP	1.526	1.630	1.630	1.625	1.621	1.615	1.614	1.595	1.592	1.615	1.590	1.590	1.599	1.604	1.590	
	B3LYPultrafine	1.526	1.630	1.630	1.625	1.621	1.615	1.614	1.595	1.592	1.615	1.591	1.590	1.600	1.604	1.590	
	B3PW91	1.531	1.639	1.639	1.636	1.631	1.625	1.624	1.603	1.600	1.624	1.599	1.599	1.607	1.613	1.599	
	mPW1PW91	1.529	1.638	1.638	1.636	1.630	1.624	1.623	1.602	1.599	1.624	1.598	1.597	1.606	1.613		
	M06-2X	1.520	1.620	1.620	1.623	1.620	1.615	1.614	1.591	1.590	1.619	1.588	1.587	1.596	1.604	1.600	
	PBEPBE	1.542	1.648			1.637	1.632	1.631	1.610	1.607	1.631	1.605	1.604	1.615	1.620	1.605	
	PBEPBEultrafine	1.542	1.648	1.648	1.643	1.637	1.632	1.631	1.611	1.607	1.631	1.605	1.604	1.615	1.620	1.605	
	PBE1PBE	1.530	1.639	1.639	1.636	1.631	1.625	1.624	1.603	1.600	1.624	1.599	1.598	1.607	1.613	1.599	
	HSEh1PBE	1.530	1.638	1.638	1.635	1.631	1.624	1.623	1.603	1.600	1.624	1.599	1.598	1.607	1.613	1.598	
	TPSSH					1.631		1.626				1.627				1.606	
	wB97X-D			1.649		1.643		1.637			1.623				1.626	1.637	1.622
	B97D3		1.656			1.647		1.640			1.618		1.616		1.626		1.615
	⋮																
Coupled Cluster	CCD	1.546	1.669	1.669	1.667	1.646	1.627	1.626	1.615	1.598	1.628	1.609	1.607	1.604	1.617	1.607	
	CCSD	1.547	1.673	1.673	1.672	1.649	1.630	1.629	1.618	1.600	1.631	1.610	1.608	1.607	1.619	1.615	
	CCSD=FULL	1.547	1.673	1.673	1.672	1.648	1.629	1.627	1.616	1.595	1.624	1.600	1.598	1.605	1.615	1.607	
	CCSD(T)	1.547	1.673	1.674	1.672	1.654	1.630	1.629	1.618	1.600	1.630	1.610	1.608	1.607	1.619	1.608	
	CCSD(T)=FULL	1.547	1.674	1.674	1.672	1.652	1.629	1.628	1.615	1.595	1.620	1.600	1.597	1.605	1.615	1.596	
		STO-3G	3-21G	3-21G*	6-31G	6-31G*	6-31G**	6-31+G**	6-311G*	6-311G**	6-31G(2df,p)	6-311+G(3df,2p)	6-311+G(3df,2pd)	TZVP	cc-pVDZ	cc-pVTZ	

General tips)

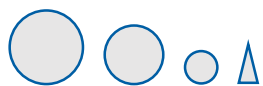
- The rightmost one with a similar name(basis sets)
- e.g.) B3LYP or CCSD(T) / aug-cc-pvqz

* Basis sets

$$\hat{H}\Psi = E\Psi$$



Gaussian-type orbital, GTO



(Basis functions)

* Dunning family) $cc-pVnZ$, $n=D,T,Q,5,\dots$

Pople) $6-31G + \{*, +, ()\}$

cc-pVDZ	cc-pVTZ	cc-pVQZ	aug-cc-pVDZ	aug-cc-pVTZ
---------	---------	---------	-------------	-------------

STO-3G	3-21G	3-21G*	6-31G	6-31G*	6-31G**
--------	-------	--------	-------	--------	---------

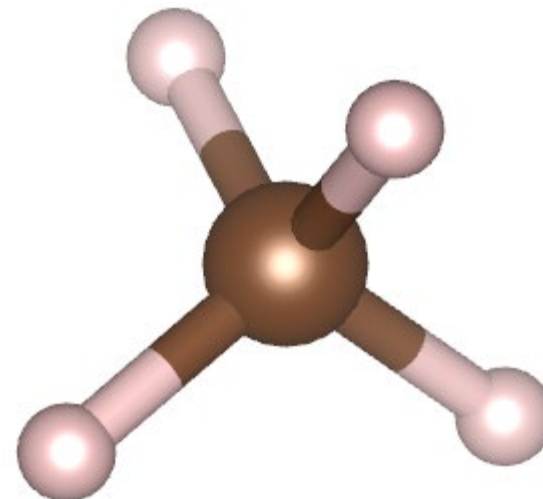
Part II. Geometry Optimization

Exercise 2. CH₃OH(Methanol) optimization

Suppose CH₃OH structure is unknown. (can not find in database.)

Strategy) Replace one H with OH in the CH₄ structure.

CH ₄	X	Y	Z
C	0.0000	0.0000	0.0000
H	0.6386	0.6386	0.6386
H	-0.6386	-0.6386	0.6386
H	-0.6386	0.6386	-0.6386
H	0.6386	-0.6386	-0.6386



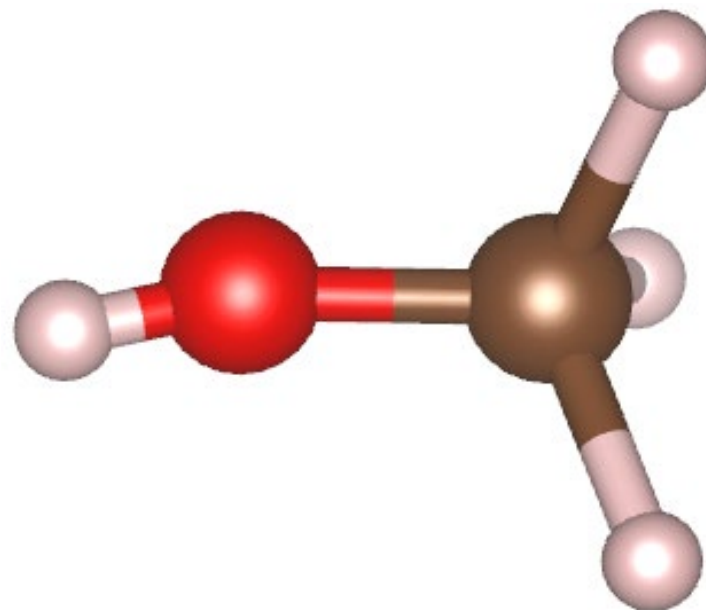
Part II. Geometry Optimization

Exercise 2. CH₃OH(Methanol) optimization

Suppose CH₃OH structure is unknown. (can not find in database.)

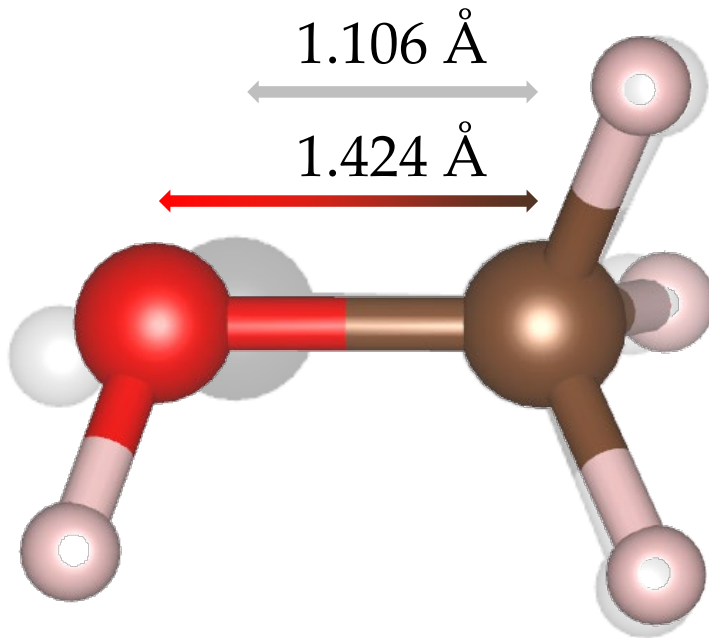
Strategy) Replace one H with OH in the CH₄ structure.

CH ₃ OH (ini. gue.)	X	Y	Z
C	0.0000	0.0000	0.0000
H	0.6386	0.6386	0.6386
H	-0.6386	-0.6386	0.6386
H	-0.6386	0.6386	-0.6386
O	0.6386	-0.6386	-0.6386
H	1.1386	-0.9386	-0.9386

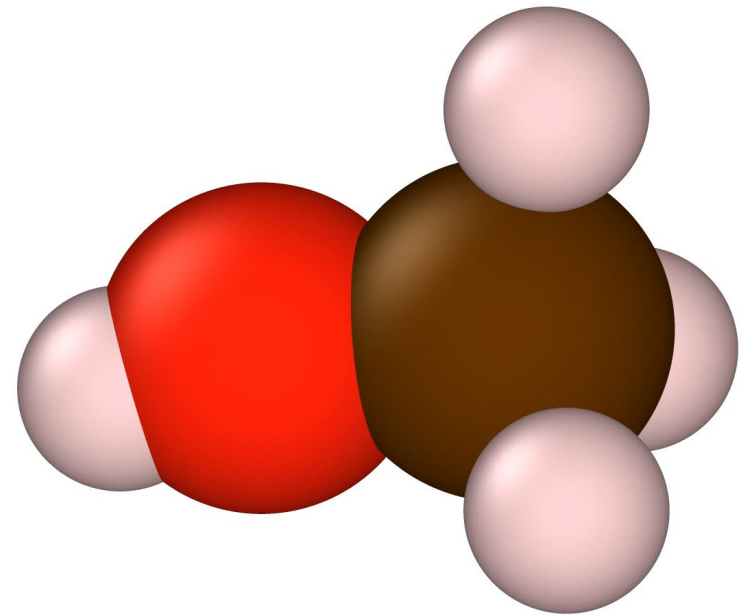


Part II. Geometry Optimization

Exercise 2. CH₃OH(Methanol) optimization



<VESTA>



<OVITO>

Part II. Geometry Optimization

Exercise 2. CH₃OH(Methanol) optimization

** Generating a reasonable initial guess structure (Maestro)

The screenshot displays the Maestro (Academic Use Only) - Scratch Project window. The interface includes a menu bar (File, Edit, Select, Workspace, Scripts, View, Window, Help) and a toolbar with various tools. The 'STYLE' menu is open, showing options like 'Color Atoms', 'Element + Custom Ligand', 'Apply Labels', 'Ribbons', and 'Surface'. A 3D model of a molecule is shown in the center, with a red 'X' overlaid on it, indicating a failed or incorrect structure. The 'ENTRY LIST' on the left shows a single entry 'str' with a blue dot next to it. The 'STRUCTURE HIERARCHY' on the bottom left shows 'Current Selection' and 'str'.

Part II. Geometry Optimization

Exercise 2. CH₃OH(Methanol) optimization

** Generating a reasonable initial guess structure (Maestro)

BUILD

3D Builder

H

Other Edits

(if needed) Add H

Select H

2 Atoms Selected

- Measure Length
- Adjust Length
- Set Length...
- Switch to Bond Menu
- Undisplay
- Style
- Expand Selection
- Focus View
- Mutate Residue
- Find Torsions
- Delete Atoms

Set (dihedral-)angle, length

Part II. Geometry Optimization

Database <Periodic system>

The screenshot shows the homepage of The Materials Project. The navigation bar at the top includes 'Apps', 'About', 'Community', 'ML', 'API', and a user profile icon. The 'Apps' dropdown menu is open, listing various tools. 'Materials Explorer' is highlighted with a red underline and a red circle containing the number '1'. A white callout box with the text 'Materials Explorer' is positioned to the right of the menu item. The main content area features the project logo and a description: 'Harnessing the power of supercomputing and state-of-the-art methods, the Materials Project provides open web-based access to computed information on known and predicted materials as well as powerful analysis tools to inspire and design novel materials.'

Part II. Geometry Optimization

Database <Periodic system>

e.g.) Bi_2Te_3 unitcell structure

The Materials Project

Home / Apps / Materials Explorer

Materials Explorer
App by Materials Project

Search for materials

Materials Bi-Te

12 materials match your search
Showing 1-12

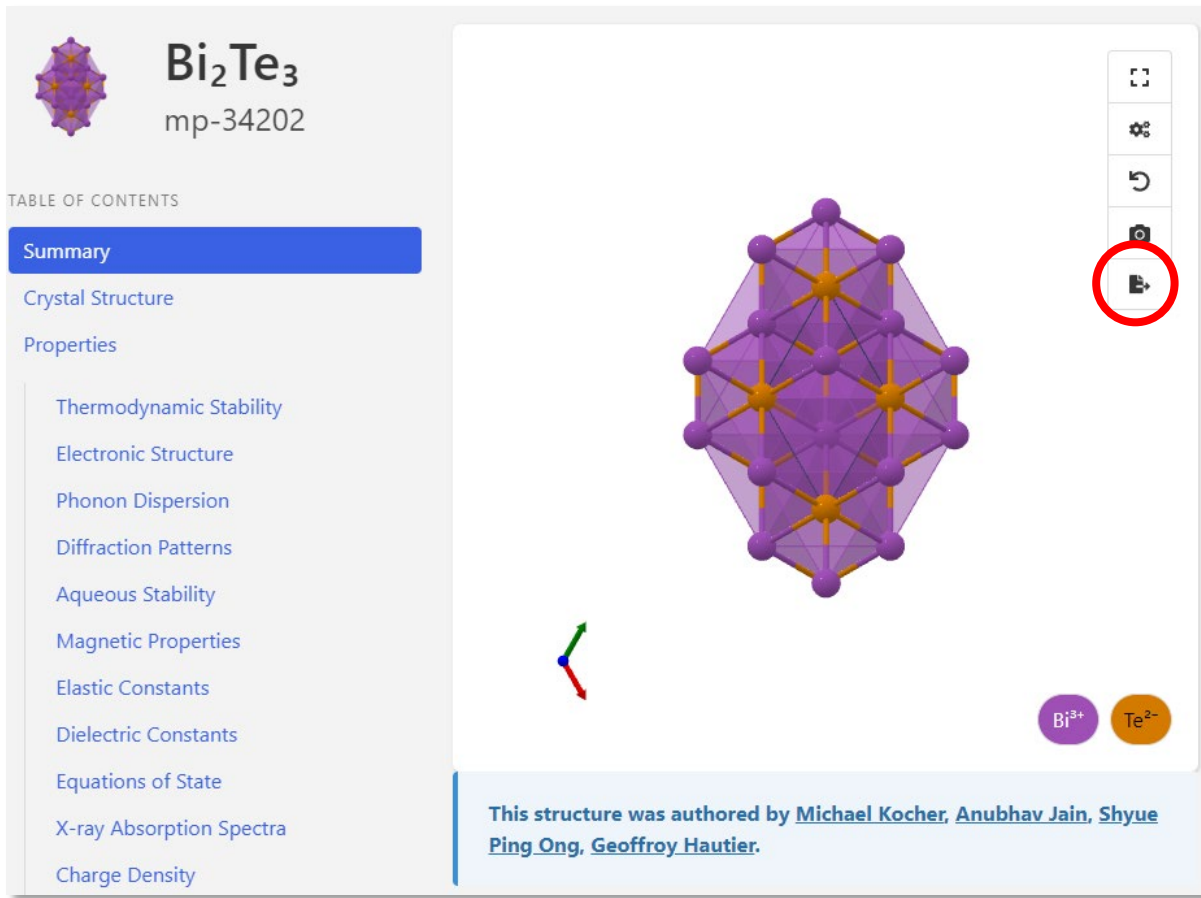
✕ Chemical System: Bi-Te

Material ID	Formula	Crystal System	Space Group Symbol
★ mp-34202	Bi_2Te_3	Trigonal	$R\bar{3}m$
mp-28229	Bi_4Te_3	Trigonal	$R\bar{3}m$
★ mp-23224	BiTe	Trigonal	$P\bar{3}m1$
mp-1214397	Bi_8Te_7	Trigonal	$P\bar{3}m1$

Part II. Geometry Optimization

Database <Periodic system>

e.g.) Bi_2Te_3 unitcell structure

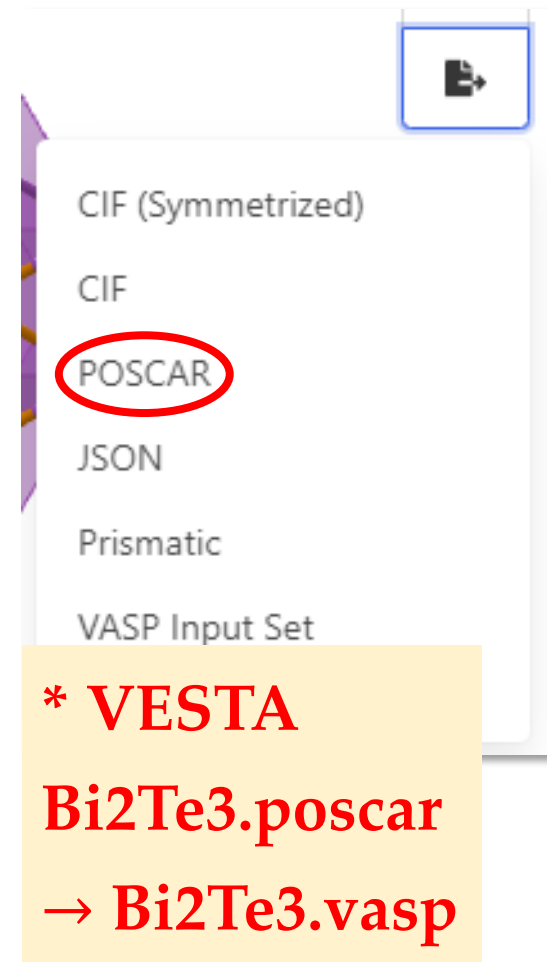


Bi₂Te₃
mp-34202

TABLE OF CONTENTS

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 - Phonon Dispersion
 - Diffraction Patterns
 - Aqueous Stability
 - Magnetic Properties
 - Elastic Constants
 - Dielectric Constants
 - Equations of State
 - X-ray Absorption Spectra
 - Charge Density

This structure was authored by [Michael Kocher](#), [Anubhav Jain](#), [Shyue Ping Ong](#), [Geoffroy Hautier](#).



- CIF (Symmetrized)
- CIF
- POSCAR**
- JSON
- Prismatic
- VASP Input Set

* **VESTA**
Bi₂Te₃.poscar
→ Bi₂Te₃.vasp

Part II. Geometry Optimization

Database <Periodic system>

e.g.) Bi_2Te_3 unitcell structure

```

Bi6 Te9
1,0
 2.2090652234007724 -3.8262132041636301
 2.2090652234007724 3.8262132041636301
 0.0000000000000000 0.0000000000000000
Bi Te
6 9
direct
 0.3333333333333333 0.6666666666666666
 0.3333333333333333 0.6666666666666667
 0.0000000000000000 0.0000000000000000
 0.0000000000000000 0.0000000000000000
 0.6666666666666666 0.3333333333333333
 0.6666666666666666 0.3333333333333335
 0.6666666666666666 0.3333333333333333
 0.0000000000000000 0.0000000000000000
 0.0000000000000000 0.0000000000000000
 0.3333333333333333 0.6666666666666666
 0.6666666666666666 0.3333333333333333
 0.6666666666666666 0.3333333333333333
 0.0000000000000000 0.0000000000000000
 0.0000000000000000 0.0000000000000000
 0.3333333333333333 0.6666666666666666
 0.6666666666666666 0.3333333333333333
 0.3333333333333333 0.6666666666666666
 0.6666666666666666 0.3333333333333333
 0.0000000000000000 0.0000000000000000
 0.3333333333333333 0.6666666666666666
 0.3333333333333333 0.6666666666666666
  
```

* POSCAR format

Comment line

Scaling factor

Lattice vectors

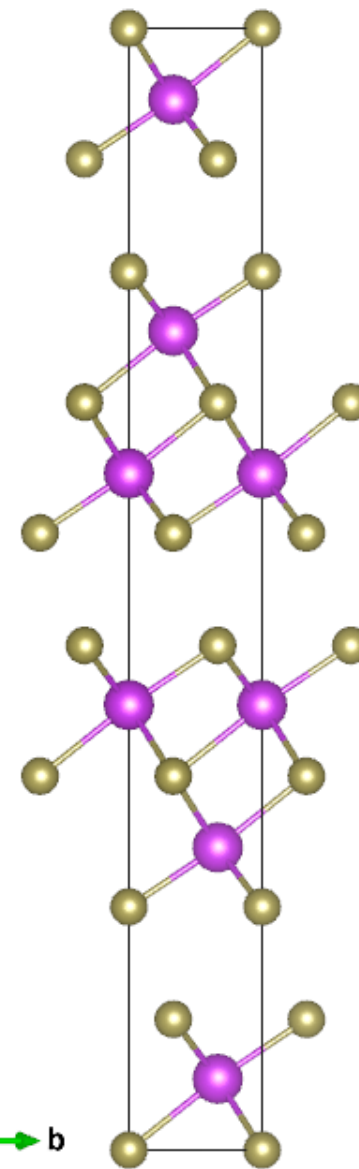
Atoms

Number of atoms

Positions

```

0.2168077399999997 Te2-
0.4498589266666669 Te2-
0.3333333333333333 Te2-
0.5501410733333331 Te2-
0.7831922600000002 Te2-
0.6666666666666666 Te2-
0.8834744066666664 Te2-
  
```



Part III. Dispersion Correction

(Grimme) DFT-D4

$$E_{total} = E_{DFA} + E_{disp} \quad E_{DFA} : \text{Part I}$$

* Lots of dispersion correction methods. (XDM, MBD, vv10, ...)

⇒ DFT-D4 linked with PySCF

```
▶ pip install dftd4
```

```
Looking in indexes: https://pypi.org/simple, https://us-python.pkg.dev/colab-wheels/public/simple/
```

```
Collecting dftd4
```

```
  Downloading dftd4-3.5.0-cp39-cp39-manylinux_2_12_x86_64.manylinux2010_x86_64.whl (15.2 MB)
```

```
----- 15.2/15.2 MB 11.0 MB/s eta 0:00:00
```

```
Requirement already satisfied: numpy in /usr/local/lib/python3.9/dist-packages (from dftd4) (1.22.4)
```

```
Requirement already satisfied: cffi in /usr/local/lib/python3.9/dist-packages (from dftd4) (1.15.1)
```

```
Requirement already satisfied: pycparser in /usr/local/lib/python3.9/dist-packages (from cffi->dftd4) (2.21)
```

```
Installing collected packages: dftd4
```

```
Successfully installed dftd4-3.5.0
```

Part III. Dispersion Correction

<Input code>

```
import dftd4.pyscf as disp
from pyscf import gto, dft
```

```
bas='ccpvdz'
```

```
xc = 'pbe'
```

```
dim = gto.M(
    atom = '''
0 0.000 0.000 0.000
H -0.635 0.628 0.391
H 0.869 0.421 0.090
O -0.277 0.479 -2.567
H -0.975 -0.048 -2.956
H -0.230 0.212 -1.603''',
    spin=0,
    charge=0,
    basis = bas,
    verbose=0)
```

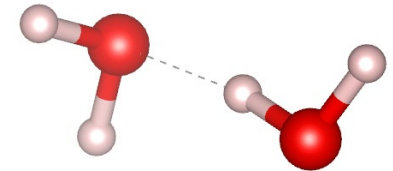
Added line

```
mydft = dft.RKS(dim)
mydft.xc = xc
e_dfa = mydft.kernel()
```

```
d4 = disp.DFTD4Dispersion(mol,xc=xc)
e_d4 = d4.kernel()[0]
```

```
e_dimer = e_dfa + e_d4
```

* Interaction energy



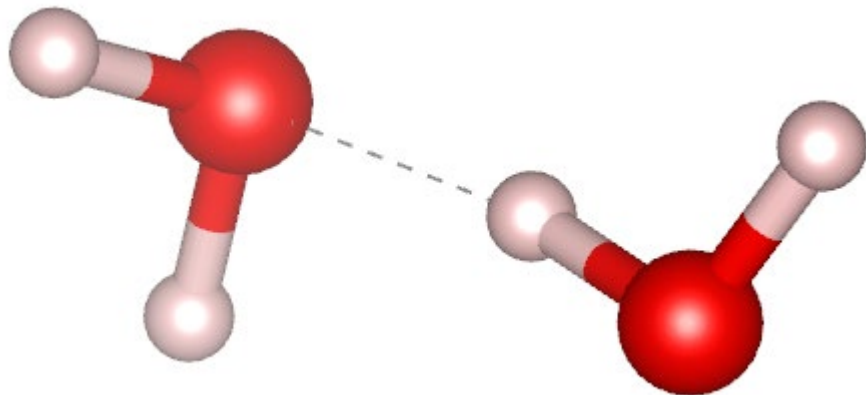
$$E_{int} = E_{dimer} - E_{mon1} - E_{mon2}$$

$$\text{Part I. } E_{sys} = E_{DFA(B3LYP, PBE...)}$$

$$\text{Part III. } E_{sys} = E_{DFA} + E_{disp}$$

Part III. Dispersion Correction

Exercise 1. Water dimer interaction energy



Ref. -3.395 kcal/mol

DLPNO-CCSD(T)-F12/aug-cc-pVQZ

	B3LYP	B3LYP-D4	PBE	PBE-D4
cc-pVDZ	-6.810	-7.501	-7.985	-8.427
cc-pVTZ	-4.469	-5.160	-5.565	-6.007
cc-pVQZ	-3.659	-4.350	-4.757	-5.199

- (D4) Independent on basis sets
- Mostly improved results (exceptions of course exist.) right here..

**PBE-D4 value corrected
(23.05.01)**

Part I. Basic DFT Calculation

- Make sure the input is well defined as the user intended.
- Comprehensive insight into {functional/basis sets} is required.

Part II. Geometry Optimization

- Initially, refer to the published database.
- If the desired structure does not exist, initial structure can be created from similar structures.

Part III. Dispersion Correction

- Almost all of the dispersion corrected DFT calculations give improved results.

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